

UNIVERSITY OF CAMBRIDGE INTERNATIONAL EXAMINATIONS General Certificate of Education Advanced Subsidiary Level and Advanced Level

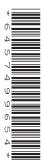
CHEMISTRY 9701/36

Advanced Practical Skills 2

October/November 2011

CONFIDENTIAL INSTRUCTIONS

Great care should be taken to ensure that any confidential information given does not reach the candidates either directly or indirectly.



The Supervisor's attention is drawn to the form on page 7 which must be completed and returned with the scripts.

If you have any problems or queries regarding these instructions, please contact CIE

by e-mail: international@cie.org.uk

by phone: +44 1223 553554 by fax: +44 1223 553558

stating the Centre number, the nature of the query and the syllabus number quoted above.

This document consists of 8 printed pages.



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Safety

Supervisors are advised to remind candidates that **all** substances in the examination should be treated with caution. Only those tests described in the question paper should be attempted. Please also see under 'Apparatus' on the use of pipette fillers, safety goggles and plastic gloves.

In accordance with COSHH (Control of Substances Hazardous to Health) Regulations, operative in the UK, a hazard appraisal of the examination has been carried out.

Attention is drawn in particular, to certain materials used in the examination. The following codes are used where relevant.

C = corrosive substance F = highly flammable substance

H = harmful or irritating substance **O** = oxidising substance

T = toxic substance N = dangerous for the environment

The attention of Supervisors is drawn to any local regulations relating to safety, first-aid and disposal of chemicals.

'Hazard Data Sheets', relating to materials used in this examination, should be available from your chemical supplier.

Before the Examination

1 Access to the question paper is NOT permitted in advance of the examination.

2 Preparation of materials

Where quantities are specified for each candidate, they are sufficient for the experiments described in the question paper to be completed.

In preparing materials, the bulk quantity for each substance should be increased by 25% as spare material should be available to cover accidental loss. More material may be supplied if requested by candidates, without penalty.

All solutions should be bulked and mixed thoroughly before use to ensure uniformity.

Every effort should be made to keep concentrations accurate to within one part in two hundred of those specified.

3 Labelling of materials

Materials must be labelled as specified in these instructions. Materials with an **FB** code number should be so labelled **without** the identities being included on the label. Where appropriate the identity of an **FB** coded chemical is given in the question paper.

4 Identity of materials

It should be noted that descriptions of solutions given in the question paper may not correspond exactly with the specifications in these instructions. The candidates must assume the descriptions given in the question paper.

5 Size of group

In view of the difficulty in preparing large quantities of solution of uniform concentration, it is recommended that the maximum number of candidates per group be 30 and that separate supplies of solutions be prepared for each group.

https://xtremepape.rs/

Apparatus

- 1 In addition to the fittings ordinarily contained in a chemical laboratory, the apparatus and materials specified below will be necessary.
- 2 Pipette fillers (or equivalent safety devices), safety goggles and disposable gloves should be used where necessary.
- 3 For each candidate
 - $2 \times 50 \,\mathrm{cm}^3$ burette (one of these should be labelled **FB 1**)
 - 2 × stand and burette clamp
 - 2 × funnel for filling burette
 - 1 × white tile
 - $2 \times 25 \, \text{cm}^3$ measuring cylinder
 - $1 \times 100 \, \text{cm}^3 \, \text{beaker}$
 - 1 × stop clock or sight of a clock with seconds display
 - 1 × Bunsen burner
 - 1 × heat proof mat
 - 1 × test-tube rack
 - 1 × test-tube holder
 - 8 × test-tubes*
 - 2 × boiling tube*
 - $3 \times \text{teat/dropping pipette}$
 - 1 × glass rod
 - $1 \times$ wash bottle containing distilled water paper towels

^{*}Candidates are expected to rinse and re-use test-tubes and boiling tubes where possible. Additional tubes should be available.

Chemicals Required

It is especially important that great care is taken that the confidential information given below does not reach the candidates either directly or indirectly. **←** © UCLES 2011

Particular requirements

hazard	label	per candidate	identity	notes (hazards given in this column refer to the raw materials)
[H] [F]	FB 1	120 cm ³	2 mol dm ⁻³ propanone	Make up $147\mathrm{cm}^3$ of propanone [H] [F] to $1\mathrm{dm}^3$ with distilled water. Keep in a stoppered bottle to avoid loss by evaporation.
Έ	FB 2	120 cm ³	3 mol dm ⁻³ hydrochloric acid	Dilute 255 cm 3 of concentrated (35% w/w; approximately 11 mol dm $^{-3}$) hydrochloric acid [C] to 1 dm 3 .
	FB 3	100 cm ³	0.005 mol dm ⁻³ iodine	To make 250 cm ³ of 0.1 moldm ⁻³ iodine, dissolve 10g of potassium iodide in 200 cm ³ of water and add 6.35g of iodine [N] [H]. To check that the iodine has dissolved, pour the solution from one beaker to another. Once all the iodine has dissolved, make up the solution to 250 cm ³ with water. Dilute the mixture 20 fold to give a concentration of 0.005 moldm ⁻³ .
Ξ	FB 4	15cm ³	1 mol dm ⁻³ sulfuric acid	See preparation instructions on page 65 of the 2011 syllabus.
[N] [E]	FB 5	15cm ³	0.1 moldm ⁻³ lead nitrate	See preparation instructions on page 65 of the 2011 syllabus.
	FB 6	15 cm ³	0.2 mol dm ⁻³ ammonium chloride and 0.3 mol dm ⁻³ sodium carbonate	Dissolve 10.7 g of freshly purchased NH ₄ Cl [H] and 85.8 g of Na ₂ CO ₃ .10H ₂ O [H] in each dm ³ of solution.

invigilator should be drawn to the fact that such an arrangement may lead to the contamination of reagents and enhance the opportunity for The reagents below should also be provided. If necessary, they may be made available from a communal supply: however, the attention of the malpractice between candidates.

	identity (hazards given in this column refer to the raw materials) 0.1 mol dm ⁻³ potassium Dissolve 19.4g of K ₂ CrO ₄ [T] [N] in each dm³ of solution. 0.5 mol dm ⁻³ sodium Dissolve 53.0g of Na ₂ CO ₃ [H] in each dm³ of solution. See identity details and preparation instructions on page 65 and 66 of the 2011 syllabus	aqueous potassium chromate(VI) aqueous sodium carbonate dilute hydrochloric acid dilute sulfuric acid aqueous ammonia aqueous sodium hydroxide 0.1 mol dm ⁻³ barium chloride or nitrate 0.05 mol dm ⁻³ silver nitra limewater	Hazard [H] [H] [H] [H] [H] [H] [H] [H
		limewater	Ξ
		0.1 moldm ⁻³ lead(II) nitra	
0.1 mold		0.05 moldm ⁻³ silver nitra	
		0.1 mol dm ⁻³ barium nitrate	Ξ
0.1 moldm ⁻³ barium nitrate 0.05 moldm ⁻³ silver nitrate 0.1 moldm ⁻³ lead(II) nitrate limewater	_	0.1 mol dm ⁻³ barium chloride or	Ε
0.1 moldm ⁻³ barium chloride or 0.1 moldm ⁻³ barium nitrate 0.05 moldm ⁻³ silver nitrate 0.1 moldm ⁻³ lead(II) nitrate		aqueous sodium hydroxide	ට
aqueous sodium hydroxide 0.1 mol dm ⁻³ barium or 0.1 mol dm ⁻³ barium nitrate 0.05 mol dm ⁻³ silver nitrate 0.1 mol dm ⁻³ lead(II) nitrate		aqueous ammonia	Ξ
aqueous ammonia aqueous sodium hydroxide 0.1 moldm ⁻³ barium or 0.1 moldm ⁻³ barium nitrate 0.05 moldm ⁻³ silver nitrate 0.1 moldm ⁻³ lead(II) nitrate limewater		dilute sulfuric acid	Ξ
dilute sulfuric acid aqueous ammonia aqueous sodium hydroxide 0.1 moldm ⁻³ barium chloride or 0.1 moldm ⁻³ barium nitrate 0.05 moldm ⁻³ silver nitrate 0.11moldm ⁻³ lead(II) nitrate limewater		dilute nitric acid	<u></u>
dilute nitric acid dilute sulfuric acid aqueous ammonia aqueous sodium hydroxide 0.1 moldm ⁻³ barium chloride or 0.1 moldm ⁻³ barium nitrate 0.05 moldm ⁻³ silver nitrate 0.05 moldm ⁻³ lead(II) nitrate limewater		dilute hydrochloric acic	Ξ
dilute hydrochloric acid dilute nitric acid aqueous ammonia aqueous sodium hydroxide 0.1 moldm ⁻³ barium chloride or 0.1 moldm ⁻³ barium nitrate 0.05 moldm ⁻³ silver nitrate 0.05 moldm ⁻³ lead(II) nitrate limewater		aqueous sodium carbonate	
aqueous sodium carbonate 0.5 mol dm ⁻³ sodium aqueous ammonia aqueous sodium hydroxide 0.1 mol dm ⁻³ barium nitrate See identity 0.05 mol dm ⁻³ silver nitrate limewater See identity		aqueous potassium chromate(VI)	[N]
aqueous potassium chromate(VI) aqueous sodium carbonate dilute hydrochloric acid dilute sulfuric acid aqueous ammonia aqueous sodium hydroxide 0.1 moldm-3 barium chloride or 0.1 moldm-3 barium nitrate 0.05 moldm-3 lead(II) nitrate limewater		label	azard

The following materials and apparatus should be available.

red and blue litmus papers, plain filter paper strips for use with dichromate(VI), aluminium foil for testing nitrate/nitrite, wooden splints and the apparatus normally used in the Centre for use with limewater in testing for carbon dioxide

Responsibilities of the Supervisor during the Examination

1 The Supervisor, or other competent chemist must, out of sight of the candidates, carry out the experiment in Question 1 and complete tables of readings on a spare copy of the question paper which should be labelled 'Supervisor's Results'.

This should be done for:

each session held and each laboratory used in that session, and each set of solutions supplied.

N.B. The question paper cover requests the candidate to fill in details of the examination session and the laboratory used for the examination.

It is essential that each packet of scripts contains a copy of the applicable Supervisor's Results as the candidates' work cannot be assessed accurately without such information.

2 The Supervisor must complete the Report Form on page 7 to show which candidates attended each session. If all candidates took the examination in one session, please indicate this on the Report Form. A copy of the Report Form must accompany each copy of the Supervisor's Results in order for the candidates' work to be assessed accurately.

The Supervisor must give details on page 8 of any particular difficulties experienced by a candidate, especially if the Examiner would be unable to discover this from the written answers.

After the Examination

Each envelope returned to Cambridge must contain the following items.

- 1 The scripts of those candidates specified on the bar code label provided.
- 2 A copy of the Supervisor's Report relevant to the candidates in 1.
- **3** A copy of the Report Form, including details of any difficulties experienced by candidates (see pages 7 and 8).
- 4 The Attendance Register.
- 5 A Seating Plan for each session/laboratory.

Failure to provide appropriate documentation in each envelope may cause candidates to be penalised.

COLOUR-BLINDNESS

With regard to colour-blindness – a minor handicap, relatively common in males – it is permissible to advise candidates to request assistance on colours of, for example precipitates and solutions (especially titration end-points). Please include with the scripts a note of the candidate numbers of such candidates.

Experience suggests that candidates who are red/green colour-blind – the most common form – do not generally have significant difficulty. Reporting such cases with the scripts removes the need for a 'Special Consideration' application for this handicap.

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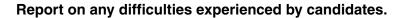
REPORT FORM

This form must be completed and sent to the Examiner in the envelope with the scripts.				
Cen	tre Number	Name of Centre		
1	Supervisor's Results			
	Please submit details of the readings obtained in Question 1 on a spare copy of the question paper clearly marked 'Supervisor's Results' and showing the Centre number and appropriate session/laboratory number .			
2	The candidate numbers of candidates attending each session were:			
	First Session	Second Session		
3	The Supervisor is required to give details overleaf of any difficulties experienced by particular candidates, giving names and candidate numbers. These should include reference to:			
	(a) any general difficulties encountered in making preparation;			
	(b) difficulties due to faulty apparatus or materials;			
	(c) accidents to apparatus or materials;			

Other cases of hardship, e.g. illness, temporary disability, should be reported direct to CIE on the normal 'Application for Special Consideration' form.

(d) assistance with respect to colour-blindness.

A plan of work benches, giving details by candidate numbers of the places occupied by the candidates for each experiment for each session, must be enclosed with the scripts.



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